# Advanced Membrane Reactors in IGCC: H<sub>2</sub> or CO<sub>2</sub> separation?

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### **Abstract**

This paper presents the results of preliminary system assessments that were conducted in order to pursue optimum operating conditions for advanced separation enhanced water-gas-shift membrane reactors. The project is entitled: "Advanced Membrane Reactors in Energy Systems" and appertains to the Global Climate & Energy Project (GCEP), which is coordinated by Stanford University. The implementation of hydrogen- and carbon dioxide-selective water-gas-shift membrane reactors in Integrated Gasification Combined Cycles was optimized with exergy analyses and compared with respect to efficiency penalties. The optimization resulted in optimum operating conditions, which serves the selection of a designated material for the development of advanced carbon dioxide-selective membrane reactors.

#### Introduction

Carbon capture and storage is widely recognized as the designated pathway towards sustainable application of fossil fuels. However, the efficiency penalty induced by carbon capture within energy conversion systems poses a threat to the economic viability of these systems. An opportunity lies in the integration of various unit operations, which reduces the efficiency penalty and consequently enhances the outlook for commercial application.

In advanced membrane reactors fuel conversion reactions such as reforming or the water-gas-shift reactions are combined with separation of one of the reaction products. Thus the fuel chemical equilibrium is shifted and conversion is increased significantly. The equilibrium shift results in reduced thermodynamic losses, when compared with conventional unit operations.

The use of coal for electric power generation by Integrated Gasification Combined Cycles (IGCC) regained interest upon the recent increase of oil and natural gas prices. The ample availability of coal which is secured for a number of centuries and its more homogeneous distribution across the globe, compared to oil and natural gas are the main factors. IGCC will play a predominant part for electricity generation in the near future. This coal conversion technology is cleaner and more efficient compared to pulverized coal boilers. Its main disadvantage remains the CO<sub>2</sub> emission, the main cause of anthropogenic global warming. Therefore IGCC with carbon dioxide capture and storage appears inevitable in pursuit of stabilization of the atmospheric carbon dioxide concentration.

Delft University of Technology and ECN cooperate in the "Advanced Membrane Reactor in Energy Systems" Project, which appertains to the Global Climate & Energy Project coordinated by Stanford University. This project pursues the development of innovative separation enhanced reactors, i.e. membrane reactors, for application in carbon-free hydrogen production or electricity generation. ECN identified hydrotalcites as designated material for carbon dioxide-selective water-gas-shift membrane reactors (WGS-MR). The work presented here is part of preliminary system assessments to identify optimum operating parameters for the carbon dioxide-selective WGS-MR.

Two advanced membrane reactor configurations were compared with respect to use in an IGCC with carbon capture. The configurations have a water-gas-shift reactor for synthesis gas conversion, which is integrated with a hydrogen- or carbon dioxide-selective membrane. We used exergy analysis to optimize each of the energy conversion systems. Then the efficiency penalties for CO2 removal were compared for the reference IGCC, for an IGCC with a carbon dioxide-selective or hydrogen selective WGS-MR and for an IGCC with a high- and low-temperature WGS reactor and carbon capture by a Selexol absorber.

## **Reference IGCC**

Coal gasification technologies can be divided in slurry- and dry-fed gasifiers. Slurry-fed gasifiers generally demonstrate lower cold-gas efficiencies than dry-fed gasifiers, due to the evaporation of the water content in the slurry. However, slurry-fed gasifiers can be operated at higher pressures when compared to dry-fed gasifiers. A significant amount of the water content within syngas obtained from a slurry-fed gasifier condenses within the acid gas removal (AGR) train. This requires a sour syngas water-gas-shift to prevent excessive downstream steam addition in order to make up for this condensation. AGR must be placed upstream of the membrane reactor, since membrane reactors are expected to demonstrate vulnerability for sulphurous impurities present in syngas. Consequently, the dry-fed Shell Coal Gasification Process (SCGP) [1] is initially selected for the implementation of advanced membrane reactors in an IGCC.

The reference IGCC that is applied during the system assessments is depicted in Figure 1. It displays the integration between the Air Separation Unit (ASU) and the gas turbine. The Shell gasifier is not refractory-lined but comprises a membrane wall that is cooled by steam generation. Further syngas cooling occurs in the gas cooler, again by steam generation. The integration of the steam flows from the gasifier and the gas cooler are omitted in all process schemes, however this steam is also expanded in the steam turbine to generate electricity. A small amount of the nitrogen produced by the ASU is applied to transport the pulverized coal and to feed it into the gasifier. The remaining nitrogen is fed to the combustion chamber of the gas turbine to lower the combustion temperature in the chamber. An H-class gas turbine combined cycle (General Electric S109H) is applied for electricity generation [2],[3]. This combined cycle comprises a single shaft lay out and a steam cycle with reheat, however these are also omitted in the graphical representations for the sake of simplicity.

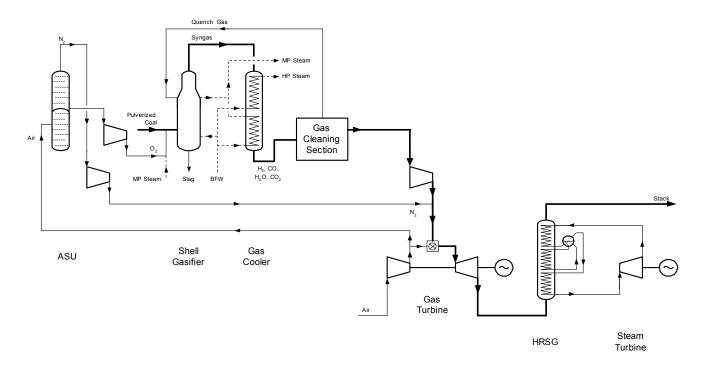


Figure 1 Reference IGCC

## **IGCC** with H<sub>2</sub>-selective WGS-Membrane Reactor

In the first case a  $H_2$ -selective WGS-MR is used downstream of the preliminary high-temperature shift reactor, as depicted in Figure 2. The WGS-MR is swept with nitrogen from the ASU in order to enhance the  $H_2$  permeation. Fuel conversion and  $H_2$  separation will not reach 100% in practice; therefore CO and  $H_2$  will still be present in the feed of the  $CO_2$  liquefaction section. The latter requires stoichiometric combustion with oxygen from the ASU in an afterburner to ensure total fuel conversion prior to  $CO_2$  liquefaction.

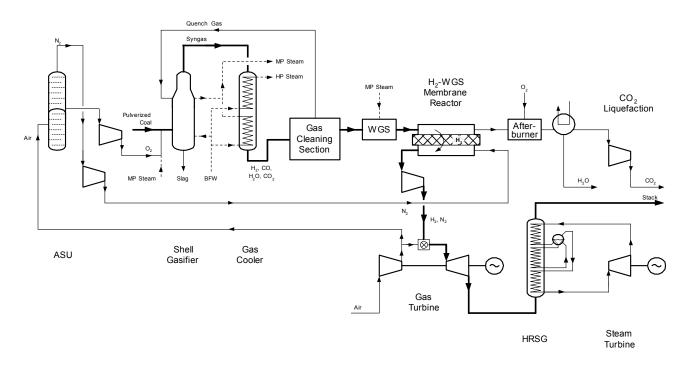


Figure 2 IGCC with H<sub>2</sub>-selective WGS-Membrane Reactor

## **IGCC** with CO<sub>2</sub>-selective WGS-Membrane Reactor

In the second case a CO<sub>2</sub>-selective membrane reactor (WGS-MR) is used downstream of the preliminary high-temperature shift reactor, as depicted in. Figure 3. The WGS-MR is swept with steam in order to enhance the CO<sub>2</sub> permeation. Fuel conversion and CO<sub>2</sub> separation will not reach 100% in practice, however this poses no problems since CO<sub>2</sub> is permeated through the membrane and any unconverted CO is fed to the combustion chamber of the gas turbine.

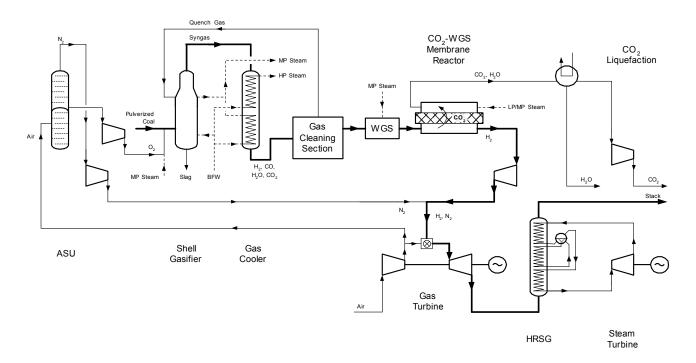


Figure 3 IGCC with CO<sub>2</sub>-selective WGS-Membrane Reactor

## IGCC with HT/LT-WGS and Selexol CO<sub>2</sub> Capture

An addition case where CO<sub>2</sub> is captured by commercially available technologies was added in order to compare the H<sub>2</sub>- and CO<sub>2</sub>-selective WGS-MR. In addition to the process depicted in Figure 1, a high-temperature (inlet: 350 °C) and low-temperature (inlet: 150 °C) WGS reactor with intermediate temperature quench are placed downstream of the AGR. CO<sub>2</sub> is captured by physical absorption in a Selexol Absorber [4].

# **Analysis**

The different cases were analyzed using AspenPlus, which was enhanced with a dedicated Fortran-based user model for the WGS-MR that was developed in-house at ECN [5]. This model comprises countercurrent dense membrane reactor configurations (amongst others) at non-isothermal conditions. Kinetic rate expressions for high-temperature WGS catalysts and temperature dependent permeation are both integrated in the model [6]. The WGS-MR cases were optimized through exergy analysis, the exergy (chemical, physical, mixing and total exergy) per flow sheet stream is obtained from the add-on tool 'Exercom'. Exergy defines the potential for power conversion of process streams with respect to a given environment [7]; exergy analysis of a process therefore identifies the opportunities to reduce losses. Exergy analyses of the two WGS-MR cases in AspenPlus was conducted by sensitivity analyses, during which the following parameters were varied:

1. Steam addition for the WGS:  $1.0 - 1.5 \text{ x } (n_{\text{CO,syngas}} - n_{\text{H2O,syngas}}) \text{ kmol/h}$ 

2. Inlet pressure of the sweep flow: 8.0 - 20.0 bar

3. Inlet temperature of the feed & sweep stream: 250 - 500 °C

4. Sweep steam flow (only CO<sub>2</sub>-selective WGS-MR): 50 - 250 t/h

The most important assumptions with respect to the system assessments are:

Coal type: "Wambo" (Australia)

Outlet pressure gasifier: 25 barOutlet pressure AGR: 20 bar

• Operation both WGS-MR: counter-current

• Permeation of the H<sub>2</sub>-selective WGS-MR: 1.0·10<sup>-6</sup> mol m<sup>-2</sup> s<sup>-1</sup> Pa<sup>-1</sup>

• Selectivity of the H<sub>2</sub>-selective WGS-MR:  $H_2/CO_2 \rightarrow \infty$ 

• Permeation of the CO<sub>2</sub>-selective WGS-MR: 1.0·10<sup>-6</sup> mol m<sup>-2</sup> s<sup>-1</sup> Pa<sup>-1</sup>

Selectivity of the CO<sub>2</sub>-selective WGS-MR: CO<sub>2</sub>/H<sub>2</sub> → ∞
 Membrane area: 30,000 m<sup>2</sup>
 CO<sub>2</sub> pressure after liquefaction: 110 bar

The assumptions regarding the  $CO_2$ -selective membranes are predominantly based on data taken from  $H_2$ -selective membranes (Pd/Ag), as  $CO_2$ -selective membranes are still under development.

## **Results and Discussion**

Table 1 displays the results of the system assessments of the WGS-MR cases, as well as the results of the reference case (no CO<sub>2</sub> capture) and the Selexol CO<sub>2</sub> capture case.

Table 1 Results Optimization

Case	Net output	Efficiency	Efficiency penalty	Carbon capture ratio	Shift steam flow	Sweep flow pressure	Inlet temp. WGS- MR	Sweep flow
	$[MW_e]$	[%]	[% abs.]	[%]	[kmol/h] <sup>1</sup>	[bar]	[°C]	[t/h]
Reference IGCC	500.0	48.9	ı	ı	1	ı	ı	-
IGCC with H <sub>2</sub> -selective WGS-MR	443.9	43.4	5.5	100	1.0	19	$260^{2}$	-
IGCC with CO <sub>2</sub> -selective WGS-MR	403.3	39.4	9.5	90	1.0	8	250	50
IGCC with HT/LT-WGS & Selexol CO <sub>2</sub> Capture	413.5	40.4	8.5	90	1.1	ı	ı	-

 $<sup>[\</sup>text{kmol/h}] \times (n_{\text{CO,syngas}} - n_{\text{H2O,syngas}})$ 

The overall efficiency for the reference case is relatively high in comparison with reported values, which is mainly to ascribe to the application of the H-class gas turbine combined cycle, as well as to the integration between the ASU and the gas turbine. The results indicate that the H<sub>2</sub>-selective WGS-MR case results in the lowest efficiency penalty, whereas the CO<sub>2</sub>-selective WGS-MR case results in the highest efficiency penalty. The Selexol case results in an absolute efficiency penalty of 8.5%, which corresponds with values reported in literature [8].

The CO<sub>2</sub>-selective WGS-MR case is less efficient than the H<sub>2</sub>-selective WGS-MR case, however it should be noted that the obtained CO<sub>2</sub> stream in the latter case is only 93 mol% pure (with N<sub>2</sub> and Argon

H<sub>2</sub>-selective WGS-MR are commonly operated at temperatures of 400 °C, changing the inlet temperature to this value results in an efficiency of 42.9%

as predominant impurities). These impurities originate from the application of  $N_2$  to feed and transport pulverized coal into the gasifier, and the combustion with 95 mol% pure oxygen obtained form the ASU in the afterburner. Furthermore, a small amount of oxygen (typically 1.5 mol%) is present in  $N_2$  obtained from the ASU, which is applied as sweep flow within the  $H_2$ -selective WGS-MR case. Combustion of this oxygen on the retentate side of the membrane was neglected during the assessments, however this could result in unacceptable high temperature gradients across the counter-current membrane configuration. Subsequently, this also results in a reduction of  $H_2$  availability for the combustion chamber of the gas turbine.

The application of steam as sweep flow within the CO<sub>2</sub>-selective WGS-MR case results in an absolute efficiency penalty of approximately 3.0%. Reduction of the steam sweep flow by operating the WGS-MR at lower retentate pressures appears mandatory to facilitate competition with the H<sub>2</sub>-selective WGS-MR. Furthermore, the CO<sub>2</sub>-selective WGS-MR allows flexibility in carbon capture ratio. This enables utility companies to speculate on CO<sub>2</sub> market fluctuations by adjustment of the carbon capture ratio.

## **Conclusions and Future Work**

The results of the system assessments require further research with respect to the issues discussed in the previous paragraph. IGCC with CO<sub>2</sub> capture through CO<sub>2</sub>-selective WGS-MR has one important disadvantage compared with the H<sub>2</sub>-selective WGS-MR, being the elevated efficiency penalty. However, advantages of the CO<sub>2</sub>-selective WGS-MR are the resulting CO<sub>2</sub> purity and the flexibility with respect to fuel conversion and CO<sub>2</sub> separation. The application of slurry-fed gasification will also be assessed in future system assessments, possibly with preliminary sour WGS upstream of the AGR.

The development and experimental evaluation of hydrotalcites as CO<sub>2</sub>-selective membrane material are performed at present. The results of these evaluations will be applied in future system assessments, including costs associated with carbon capture for the presented cases.

# **Acknowledgement**

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